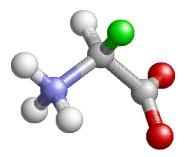
## CS612 - Algorithms in Bioinformatics

Protein Structure

February 19, 2025

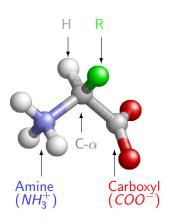
#### Introduction to Protein Structure

A protein is a linear chain of organic molecular building blocks called amino acids.

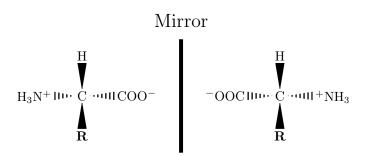


#### Introduction to Protein Structure

- Amine  $(NH_3^+)$ , carboxyl  $(COO^-)$ , C- $\alpha$  and the hydrogen attached to it are called backbone.
- All amino acids have the same backbone.
- R (residue) can be anything...
- It is called the side chain, and is different for different amino acids.
- In nature there are 20 amino acids.



#### D vs. L Amino Acids



- The dashed lines are pointing away from you, and the bold lines are pointing towards you.
- Amino acids in nature are L (the left side of the image).
- D Amino acids (right side) can be synthesized, but they nearly never exist in nature.



#### Amino Acid Names

Name	3-letter	1-letter	Side chain	Name	3-letter	1-letter	Side chain
Glycine	GLY	G		Glutamine	GLN	Q	
Alanine	ALA	Α	800	Asparagine	ASN	N	6
Phenylalanine	PHE	F		Lysine	LYS	К	E E E
Leucine	LEU	L		Arginine	ARG	R	E E CO
Isoleucine	ILE	1	688	Serine	SER	S	8.8
Valine	VAL	V	333	Threonine	THR	т	&- <del>56</del>
Proline	PRO	Р		Tyrosine	TYR	Y	
Methionine	MET	М	£ 66	Glutamic acid	GLU	E	
Tryptophan	TRP	W		Aspartic acid	ASP	D	<b>E</b>
Histidine	HIS	н	8-62	Cysteine	CYS	С	<b>&amp;</b>

# Hydrophobic Amino Acids

Name	3-letter	1-letter	Side chain	Name	3-letter	1-letter	Side chain
Glycine	GLY	G		Glutamine	GLN	Q	E de
Alanine	ALA	Α	80	Asparagine	ASN	N	
Phenylalanine	PHE	F	- A	Lysine	LYS	K	
Leucine	LEU	L		Arginine	ARG	R	E E G
Isoleucine	ILE	1	6880	Serine	SER	S	8.8
Valine	VAL	V	رگورگان	Threonine	THR	Т	& & & & & & & & & & & & & & & & & & &
Proline	PRO	Р		Tyrosine	TYR	Y	644
Methionine	MET	М	£ 6	Glutamic acid	GLU	E	8-8
Tryptophan	TRP	W		Aspartic acid	ASP	D	6
Histidine	HIS	Н	8-62	Cysteine	CYS	С	800

## Polar Amino Acids

Name	3-letter	1-letter	Side chain	Name	3-letter	1-letter	Side chain
Glycine	GLY	G		Glutamine	GLN	Q	
Alanine	ALA	Α	80	Asparagine	ASN	N	
Phenylalanine	PHE	F	900 90 00 00	Lysine	LYS	K	E E
Leucine	LEU	L		Arginine	ARG	R	E E CO
Isoleucine	ILE	1	2680	Serine	SER	S	8
Valine	VAL	V	3,3	Threonine	THR	т	& & &
Proline	PRO	Р	algo.	Tyrosine	TYR	Υ	
Methionine	MET	М	and the same of th	Glutamic acid	GLU	Е	8-6
Tryptophan	TRP	W		Aspartic acid	ASP	D	6
Histidine	HIS	Н	8-6X°	Cysteine	CYS	С	800

## Charged Amino Acids

Name	3-letter	1-letter	Side chain	Name	3-letter	1-letter	Side chain	_
Glycine	GLY	G	•	Glutamine	GLN	Q	E in	
Alanine	ALA	Α	80	Asparagine	ASN	N		
Phenylalanine	PHE	F		Lysine	LYS	K	666	Positive charge
Leucine	LEU	L		Arginine	ARG	R	E E CO	, and the second
Isoleucine	ILE	ı	288	Serine	SER	S	8.8	
Valine	VAL	V	رگر <sub>و</sub> گر هــــــــــــــــــــــــــــــــــــ	Threonine	THR	Т	& <del>%</del>	
Proline	PRO	Р	algho	Tyrosine	TYR	Υ		
Methionine	MET	М	400 m	Glutamic acid	GLU	E	<b>3-4</b>	Negative charge
Tryptophan	TRP	W		Aspartic acid	ASP	D		
Histidine	HIS	Н	8-4X	Cysteine	CYS	С		

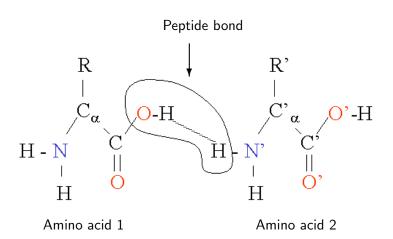
#### **Aromatic Amino Acids**

Name		3-letter	1-letter	Side chain	Name	3-letter	1-letter	Side chain
Glycir	ie	GLY	G		Glutamine	GLN	Q	E de
Alaniı	ne	ALA	Α	80	Asparagine	ASN	N	
Pheny	/lalanine	PHE	F	900 8 0 0	Lysine	LYS	K	
Leuci	ne	LEU	L		Arginine	ARG	R	E E GO
Isoleu	cine	ILE	1	2660	Serine	SER	S	8
Valine	:	VAL	V	رگر <sub>و</sub> گر م	Threonine	THR	Т	& <del>%</del>
Prolin	e	PRO	Р	algo.	Tyrosine	TYR	Y	
Methi	onine	MET	М	60 m	Glutamic acid	GLU	E	<b>%</b>
Trypt	ophan	TRP	W		Aspartic acid	ASP	D	<b>E</b>
Histid	ine	HIS	Н	8-6X°	Cysteine	CYS	С	800

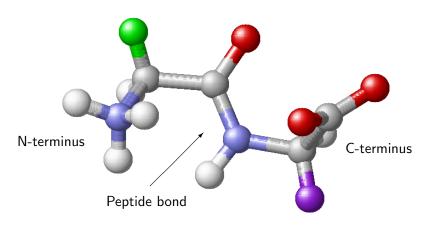
#### Formation of Proteins

- During the translation of a gene into a protein, the protein is formed by the sequential joining of amino acids end-to-end to form a long chain-like molecule (polymer).
- A polymer of amino acids is often referred to as a polypeptide.
- The genome is capable of coding for 20 different amino acids whose chemical properties. depend on the composition of their side chains ("R").
- Thus, to a first approximation, a protein is a sequence of these amino acids.
- This sequence is called the primary structure of the protein.

#### Formation of Proteins

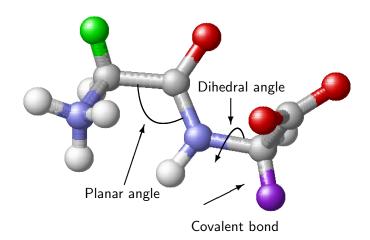


# Polymerization of Amino Acids

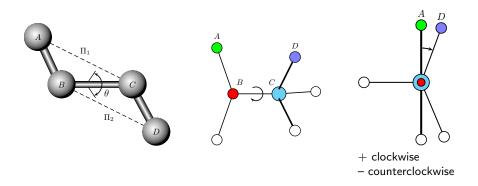


Peptide - amino acids polymerized in chain

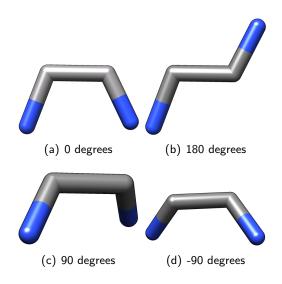
#### Interactions Between Atoms



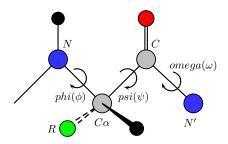
# Dihedral Angles



## Dihedral Angles



## **Backbone Dihedral Angles**

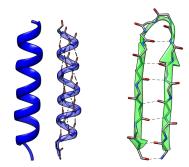


- $\phi$  is defined by  $C_{i-1}$ , N,C- $\alpha$ ,C rotation about the N C- $\alpha$  axis.
- It is undefined for the first amino acid.
- $\psi$  is defined by  $N_i$ , C- $\alpha$ , C,  $N_{i+1}$  rotation about the C- $\alpha$  C axis.
- It is undefined for the last amino acid.
- $\bullet$   $\omega$  is always 180 degrees.



# Protein Secondary Structures

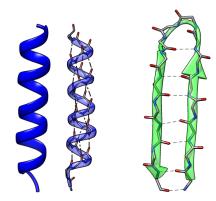
- In order to work properly, a protein must fold to form a specific three-dimensional shape called native conformation/structure.
- Secondary structure refers to folding in a small part of the protein that forms a characteristic shape.
- The most common secondary structure elements are  $\alpha$ -helices and  $\beta$ -sheets.



Left:  $\alpha$  helix. Right:  $\beta$  sheet

# Secondary Structure Elements

- Repeating values of  $\phi$  and  $\psi$  along the peptide chain result in regular structures.
- These structures are stabilized by interactions along atoms on the chain.

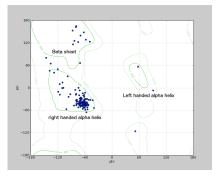


Left:  $\alpha$  helix. Right:  $\beta$  sheet

#### $\alpha$ Helices

 $\alpha$  helices are characterized by repeating values of  $\phi$  around -57 and  $\psi$  around -47.

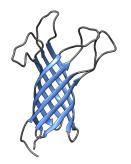


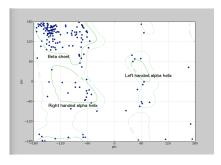


Ramachandran plot: A scatterplot showing the  $\phi$  and  $\psi$  values of amino acids. Areas in green are "allowed" (energetically favorable).

### $\beta$ Strands

 $\beta$  strands are characterized by repeating values of  $\phi$  around -110 – -140 and  $\psi$  around 110 – 135.

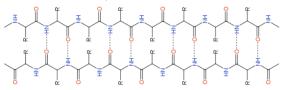




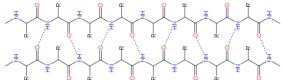
These relatively extended strands interact to form  $\beta$  sheets.

## Parallel and Anti-Parallel $\beta$ Sheets

#### Antiparallel beta-sheet

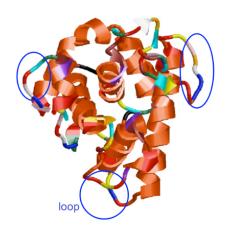


#### parallel beta-sheet



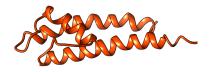
## Coils and Loops

- The sections of the peptide chain that link the  $\alpha$ -helices and  $\beta$ -sheets are referred to as turns and loops
- Other secondary substructure classifications exist, but are rarely seen in practice
- Sub-units that do not fit into any other classification are known as random coils



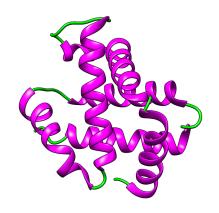
#### Protein 3-D Structure

- The 3-dimensional fold of a protein is called a tertiary structure.
- Many proteins consist of more than one polypeptide folded together.
- The spatial relationship between these separate polypeptide chains is called the quaternary structure.

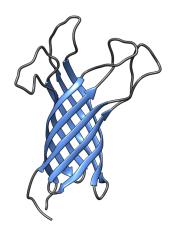




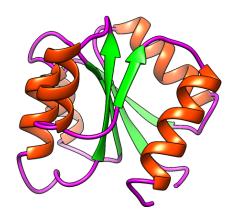
# Protein Folds – Mainly $\alpha$



# Protein Folds – Mainly $\beta$



# Protein Folds – $\alpha/\beta$

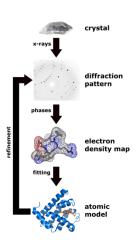


#### Protein Structure Determination

- How does a protein know its fold?
- It is believed that all the information is encoded in its primary structure (amino acid sequence).
- Yet, no algorithm exists as of today to successfully predict this structure – the protein folding problem.
- There are experimental methods to determine protein structure.

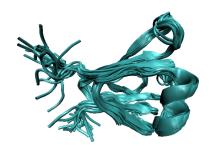
## X-ray Crystallography

- Crystallize the protein.
- Pass an X-ray to create a diffraction pattern.
- Reconstruct atomic model from electron density map.



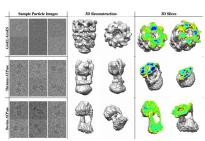
# NMR (Nuclear Magnetic Resonance) Spectroscopy

- Magnetic field is applied to a solution containing the protein.
- Atomic nuclei are aligned by the field.
- When unaligned, the nuclei give off a typical signal.
- Inter-atomic distances can be inferred.
- The 3-D structure can be modeled.
- Done in solutions atoms are free to move.
- Usually produces an ensemble of structures.



# Cryo-Electron Microscopy (Cryo-EM

- Samples are frozen to cryo-temperatures (of liquid nitrogen)
- EM is used to obtain an image.
- Multiple 2D images are used to construct a 3D image.
- Especially useful for very large macromolecules (entire viruses, ribosomes...)
- Started at 1nm resolution, nowadays approaching 2-3Å.



http://blogs.sciencemag.org/